## A New Reagent for Oxidation of Alcohols to Ketones in Neutral Solution at Room Temperature

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Incidentally to an account of the photochemical addition of alkyl ethers to diethyl azodicarboxylate1 we mentioned that the ester oxidizes cyclohexanol to cyclohexanone in the dark, though only in low yield. In fact we had carried out many experiments on a variety of alcohols under various conditions in an attempt to develop a useful preparative method of oxidation, but could not achieve yields above 20 to 30% (except for n-butanol 70%).

However, Yoneda, Suzuki, and Nitta<sup>2</sup> have just reported the oxidation of several alcohols to aldehydes and ketones in yields varying from 55 to 87% under conditions that we had already tried with little success (e.g., in boiling benzene). We have now repeated the work, and under the conditions described2 we still find yields of only half or less of Yoneda's. No significant improvement resulted from the use of samples of azo-ester prepared by different procedures or those to which possible impurities or potential catalysts had been added.

The more reactive azo-compound, 4-phenyl-1,2,4-triazoline-3,5-dione,3,4 oxidises alcohols to aldehydes or ketones at room temperature over a few hours in high yield. Dry benzene is a convenient solvent, from which the phenylurazole separates during the reaction. Some examples of the method with equimolar quantities of azocompound and alcohol are given in the Table.

## TABLE

		Yield (%)*
Alcohol	Ketone	Unreacted alcohol
Benzyl alcohol	 78†	17
Benzhydrol	 90	
Cyclopentanol	 62	34
Cyclohexanol	 84	12
4-t-Butylcyclohexanol	 75	22

<sup>\*</sup> Measured by gas chromatography, except benzophenone, which was estimated as 2,4-dinitrophenylhydrazone.

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<sup>†</sup> Benzaldehyde.

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<sup>&</sup>lt;sup>4</sup> R. C. Cookson, S. S. H. Gilani, and I. D. R. Stevens, Tetrahedron Letters, 1962, 615.